New Routes for Novel Pyrazolo[3,4-*b*][1,6]naphthyridine, Pyrazolo[3,4-*b*]pyridine and Pyrazolo[3,4:2,3]pyrido[6,1-*a*]benzimidazole Derivatives

Fawi M. Abd El Latif,* Magda A. Barsy, Eman A. Elrady and M. Hassan

Department of Chemistry, Aswan Faculty of Science, Aswan, Egypt

Pyrazolo[3,4-*b*][1,6]naphthyridine, pyrazolo[3,4-*b*]pyridine and pyrazolo[3,4:2,3]pyrido[6,1-*a*]benzimidazole derivatives are synthesized starting from the isomeric 3-substituted 5-chloro-1-phenylpyrazole-4-carbaldehyde.

Diverse biological activities have been reported for polyfunctional substituted pyrazoles.⁶ The condensation of hydrazines with α,β -unsaturated aldehydes, ketones and nitriles is regarded as one of the most useful methods for pyrazoline synthesis.⁹ Here, we report the effect of catalyst, reaction times and substituents on the reactivity of 5-chloro-3-substituted-1-phenylpyrazole-4-carbaldehyde 1 which has attracted attention as a precursor.¹¹

Thus, compound 1a (R = Ph) condensed readily with malononitrile 2a in refluxing ethanol to afford 3a in 70% yield. By contrast, 1b (R = Me) was recovered virtually unreacted when refluxed with 2a in ethanol, however, ylidene 3b was formed when the reaction was carried out in the presence of piperidine either at room temperature



*To receive any correspondence.

1999, 696–697 J. Chem. Research (M), 1999, 2954–2974

J. Chem. Research (S),

or at reflux for 2–4 h. Alternatively, **1b** reacted with **2a** in ethanolic piperidine at reflux for 15 h to yield a product formulated as a pyrazolo[3,4-*b*][1,6]naphthyridine derivative **4** (Scheme 1). It was thus assumed that **1b** condensed with malononitrile **2a** to give **3b**, which readily reacted with the malononitrile dimer **10** which formed *in situ*, to yield an intermediate Michael adduct I which cyclized into intermediate II which finally loses malononitrile¹⁵ and HCl to give **4** (Chart 1).

On the other hand, 1a,b reacted smoothly with ethyl cyanoacetate 2b in ethanolic piperidine at reflux to afford the acrylate derivatives 5a,b. However, the methylidene 6 was isolated in 42% yield when the reaction was carried out in the absence of piperidine. 1a reacted with cyanoacetamide 7a and/or cyanothioacetamide 7b in ethanolic piperidine to yield acyclic buta-1,3-diene derivatives 8a,b. The chemical structures of 8a,b were established based on IR and ¹NMR spectroscopy, MS and elemental analysis. By contrast, 1b reacted with 7a,b under the same reaction conditions to afford 1,3-diphenylpyrazolo[3,4-*b*]pyridine derivatives 9a,b. The chemical structure of 9a was confirmed by IR, ¹H and ¹³C NMR, MS and elemental analysis.

Compound 1a condensed with malononitrile dimer 10 in ethanolic piperidine to yield a product formulated as 11. The formation of 11 is believed to proceed *via* elimination of one molecule of malononitrile from the initial butadiene intermediate, which recombined once more¹⁵ to yield the final product 11. On the other hand, 1b readily condensed with 10 under the same reaction conditions to give the acyclic buta-1,3-diene derivative 12. The chemical structures of 11 and 12 were confirmed based on elemental and spectral data (Scheme 1).

Compound **1a** reacted with 2-cyanomethylbenzimidazole **13** in ethanolic piperidine at reflux to yield the new 1,3-diphenylpyrazolo[3,4:2,3]pyrido[6,1-*a*]benzimidazole **15**





Scheme 2

as the only isolable product as confirmed by IR, ¹H and ¹³C NMR, MS and elemental analysis. However, intermediate 14 was isolated in 74% yield in the absence of piperidine. Furthermore, 1a,b reacted directly with cyanoacetohydrazide 16 in ethanolic piperidine at reflux to yield the dimeric azines 17a,b which were identified by comparison with authentic samples.¹¹ However, **1b** was recovered unreacted with 16 in the absence of piperidine, while 1a afforded 19, which is assumed to be formed via initial formation of the carbohydrazide intermediate 18. The IR spectrum of 19 showed no indication of a cyano function. Finally, 1a reacted with benzoic acid hydrazide 20 in ethanolic piperidine at reflux to yield 22 (25%). Treatment of the filtrate yielded 21 in 69% yield, with elemental and spectral data consistent with the proposed structures (Scheme 2).

Techniques used: IR, ¹H and ¹³C NMR, mass spectrometry

References: 24

Schemes: 3

Charts: 3

Received, 11th August 1999; Accepted, 9th September 1999 Paper E/9/06535K

References cited in this synopsis

- 6 T. Shinichi, M. Koichi, H. Yumi, W. Katsuaki, S. Shinzaburo, O. Tooru and I. Akimi, Jpn. Pat., JP 06, 184, 114 (Chem. Abstr., 1995, **122**, 105875e). R. H. Wiley and C. H. Jarboe, *The Chemistry of Heterocyclic*
- 9 Compounds, Interscience Publishers, New York, 1967, vol. 22, part 2, p. 183.
- 11 S. A. Shiba, N. M. S. Harb, M. A. Hassan, M. A. El-Kassaby and M. M. K. Abou-El Regal, Indian J. Chem., Sect. B: Org. Chem. Incl. Med. Chem., 1996, **35**, 426. M. H. Elnagdi, M. R. H. Elmoghayar and A. H. Elghandour,
- 15 Sulfur Lett., 1989, 9, 109.